

# Kinetics and Thermodynamics of $\beta$ 2-Microglobulin Binding to the $\alpha$ 3 Domain of Major Histocompatibility Complex Class I Heavy Chain<sup>†</sup>

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**ABSTRACT:** The major histocompatibility complex (MHC) class I molecule plays a crucial role in cytotoxic lymphocyte function. Functional class I MHC exists as a heterotrimer consisting of the MHC class I heavy chain, an antigenic peptide fragment, and  $\beta$ 2-microglobulin ( $\beta$ 2m).  $\beta$ 2m has been previously shown to play an important role in the folding of the MHC heavy chain without continued  $\beta$ 2m association with the MHC complex. Therefore,  $\beta$ 2m is both a structural component of the MHC complex and a chaperone-like molecule for MHC folding. In this study we provide data supporting a model in which the chaperone-like role of  $\beta$ 2m is dependent on initial binding to only one of the two  $\beta$ 2m interfaces with class I heavy chain.  $\beta$ 2-Microglobulin binding to an isolated  $\alpha$ 3 domain of the class I MHC heavy chain accurately models the biochemistry and thermodynamics of  $\beta$ 2m-driven refolding. Our results explain a 1000-fold discrepancy between  $\beta$ 2m binding and refolding of MHC1. The biochemical study of the individual domains of complex molecules is an important strategy for understanding their dynamic structure and multiple functions.

The major histocompatibility complex class I (MHC1)<sup>1</sup> molecule and antigenic peptide are recognized by CD8+ cytotoxic T-lymphocytes (CTL) in CTL activation and lysis of targets (1). The heavy chain of the MHC class I molecule can interact noncovalently with a number of other molecules in the formation of a CTL activating complex. These include the MHC class I light chain or  $\beta$ 2m, the antigenic peptide fragment, the T-cell receptor (TCR), and the CD8 molecule (2). The specificity of the CTL response resides in the selective MHC1 binding of specific antigenic peptide fragments and in the TCR recognition of these antigenic peptides and MHC1 (3, 4). The MHC1 contact surface for TCR and peptide binding is formed by the  $\alpha$ 1 and  $\alpha$ 2 domains of the three-domain MHC class I heavy chain (2, 5–7).

$\beta$ 2m, the nonpolymorphic component of the MHC1 complex, has been shown by X-ray crystallography to interact

with the immunoglobulin-like  $\alpha$ 3 domain as well as the  $\alpha$ 1 $\alpha$ 2 domains of the MHC1 heavy chain (8). Mutations in the  $\alpha$ 1 (9, 10) or  $\alpha$ 3 (11) domains of MHC1 lead to changes in  $\beta$ 2m binding. These studies demonstrate that the functional interaction of the MHC1 heavy chain with  $\beta$ 2m occurs at multiple surfaces on different domains.

In the absence of  $\beta$ 2m, most MHC1 molecules are not expressed efficiently on the surface of cells (12, 13). Although some MHC1 molecules, such as H-2L<sup>d</sup> and H-2D<sup>b</sup>, are transported to the cell surface without  $\beta$ 2m, they have diminished levels of expression (14, 15). This decreased MHC1 expression is not simply due to an export requirement for fully assembled MHC1 complexes. Transfection of  $\beta$ 2m-negative cells with ER-retained  $\beta$ 2m was able to salvage MHC1 cell surface expression (16). MHC1 folded in the presence of  $\beta$ 2m was exported to the cell surface without bound  $\beta$ 2m. A molecule, such as  $\beta$ 2m, which promotes protein folding through a transient interaction fits the definition of a chaperone (17). Therefore,  $\beta$ 2m plays two roles in MHC1: first, as a structural subunit of the assembled complex and, second, as a chaperone for the folding of the MHC1 heavy chain. A possible mechanism for  $\beta$ 2m as a chaperone is facilitation of the interaction of MHC1 heavy chain with other chaperones, such as calreticulin, tapasin, TAP, and Erp57 (18, 19). However, since  $\beta$ 2m has been shown to promote stabilization or refolding of MHC1 on the cell surface in the absence of ER chaperones (20, 21), it is likely that  $\beta$ 2m also has a direct effect on MHC1 folding.

Although micromolar concentrations of high-affinity peptides can fold MHC1 in the absence of  $\beta$ 2m, these same

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<sup>1</sup> Abbreviations: MHC, major histocompatibility complex; MHC1, major histocompatibility class I molecules; CTL, cytotoxic T-lymphocyte; TCR, T-cell receptor; SPR, surface plasmon resonance; HBS-EP, HEPES-buffered saline with 3 mM EDTA and 0.005% polysorbate 20; h $\beta$ 2m, human  $\beta$ 2-microglobulin; m $\beta$ 2m, murine  $\beta$ 2-microglobulin;  $\alpha$ 3 par, H-2D<sup>d</sup>  $\alpha$ 3 domain fusion protein;  $\alpha$ 3 227m, H-2D<sup>d</sup>  $\alpha$ 3 domain fusion protein with a mutation at position 227.

peptides can stabilize MHC1 folded with  $\beta 2m$  at significantly lower concentrations (22, 23). Therefore, with physiologic concentrations of high-affinity peptides or any concentration of lower affinity peptides,  $\beta 2m$  is limiting for the folding of MHC1 molecules.

The two roles of  $\beta 2m$ , as structural subunit and chaperone, do not have the same dependence on  $\beta 2m$  concentration.  $\beta 2m$  binds to MHC1 heavy chain with an equilibrium dissociation constant ( $K_d$ ) in the nanomolar range (24–27) while it refolds or stabilizes cell surface MHC1 at micromolar concentrations (20, 21, 28). We have previously demonstrated that human  $\beta 2m$  ( $h\beta 2m$ ) binds the isolated  $\alpha 3$  domain of the MHC1 heavy chain with a  $K_d$  in the micromolar range (29). This suggested that  $\beta 2m$  folding of the complete MHC1 heavy chain may be nucleated by a  $\beta 2m$ – $\alpha 3$  interaction. Evidence with mutant cell lines (30) and assembly of newly translated molecules (31) suggest that the  $\alpha 3$  domain is folded before the  $\alpha 1\alpha 2$  domains and this folding can occur without  $\beta 2m$  and peptide. The heavy chain  $\alpha 3$  domain would then be available for initial binding of  $\beta 2m$ , which would facilitate folding of the  $\alpha 1\alpha 2$  domains of the heavy chain. The  $\beta 2m$ -folded  $\alpha 1\alpha 2$  domains would contribute to the heavy chain interaction with  $\beta 2m$  and increase the likelihood of peptide binding in normal cells. Bound peptide would further stabilize the  $\alpha 1\alpha 2$  domains of the heavy chain and increase the affinity of the folded MHC1 heavy chain for  $\beta 2m$  to the nanomolar range.

To evaluate this model, we have further characterized the binding of  $\beta 2m$  to  $\alpha 3$  using surface plasmon resonance (SPR). We studied the binding of murine  $\beta 2m$  ( $m\beta 2m$ ) in addition to human  $\beta 2m$  and determined the temperature dependence of the interaction. The species and temperature dependence of  $\beta 2m$  binding to  $\alpha 3$  and  $\beta 2m$  refolding of MHC1 heavy chain are very similar. The chaperone-like effects of  $\beta 2m$  can be explained by a  $\beta 2m$  interaction with the  $\alpha 3$  domain of a heavy chain with unfolded  $\alpha 1\alpha 2$  domains.

## EXPERIMENTAL PROCEDURES

**$\alpha 3$  Domain Protein Expression and Purification.** The H-2D<sup>d</sup>  $\alpha 3$  domain ( $\alpha 3$  par) sequence was generated by PCR amplification of an H-D<sup>d</sup> cDNA as described previously with the 5' primer ACTCCATGGCAACAGATCCCCAAAGGCC and the 3' primer GATGAATTCGACCCGGAAGGAGGAGTTC (29). The  $\alpha 3$  domain sequence was inserted between the *Nco*I and *Eco*RI sites of a modified pET21d vector (Novagen, Madison, WI). The vector was modified by ligating a synthetic oligonucleotide (GAGGAATTCTGGAATTCGCAAGCTGTACATGCTGCACACGCTGAAATTAACGAAGCAGGAAGAGCACTCGAGCAC) between the *Eco*RI and *Xho*I sites of the pET21d bacterial expression vector. The completed construct was sequenced and found to be correct. The resulting construct, when transfected and induced, generated a 15 kDa fusion protein consisting of the H-D<sup>d</sup>  $\alpha 3$  domain fused to a vector-expressed sequence, a 17 amino acid peptide sequence from ovalbumin, and a polyhistidine tag.

Expression constructs were transfected into BL21(DE3) bacteria (Novagen, Madison, WI). Four hundred milliliter cultures of transfected bacteria in LB broth with 200  $\mu$ g/mL carbenicillin (SIGMA, St. Louis, MO) were grown to an

absorbance of 0.6 at 600 nm. The cultures were then induced with 1 mM IPTG (Sigma), and the cells were harvested by centrifugation after 3 h of induction at 37 °C or overnight induction at 28 °C. The bacteria were washed with PBS, and the pellet was frozen at –70 °C. The frozen pellet was thawed in 0.5 M NaCl/10 mM Tris, pH 8.0, with 1 mg/mL lysozyme (Sigma). After addition of imidazole (Sigma) to a concentration of 5 mM and Triton X-100 (Boehringer Mannheim, Indianapolis, IN) to a concentration of 1%, the samples were sonicated in a Brinkmann homogenizer (Brinkmann, Westbury, CT) for 3  $\times$  30 s at a setting of 4. The homogenate was treated with ~500 units of Benzonase (Sigma) in the presence of 5 mM MgCl. Inclusion bodies were pelleted by spinning at 15000g, and the soluble fraction was loaded onto buffer equilibrated NTA-Ni-resin (Qiagen, Santa Clarita, CA) for 1 h at 4 °C. The loaded NTA-Ni-resin was washed three times with 0.5 M NaCl/10mM Tris, pH 8.0/1% Triton X-100/5 mM imidazole buffer and then three times with 0.5 M NaCl/10 mM Tris, pH 8.0, buffer. The fusion protein was eluted with high-concentration imidazole (150–500 mM). In some cases, protein eluted with 150 mM imidazole was further purified by size exclusion chromatography. The eluted protein was dialyzed into HEPES-buffered saline (HBS) with 3 mM EDTA, and the protein concentration was measured by 280 nm absorbance. The extinction coefficient at 280 nm was calculated from the primary amino acid sequence. For experiments using the BIAcore, the buffer was adjusted with 10% polysorbate 20 (Biacore AB, Uppsala, Sweden) to 0.005% polysorbate 20 (HBS-EP buffer). For generation of the E227K mutant  $\alpha 3$  domain ( $\alpha 3$  227m), an H-D<sup>d</sup> cDNA with the E227K mutation was used as the template for generating the mutant  $\alpha 3$  insert. The completed construct was sequenced and purified in the same manner as the parental  $\alpha 3$  construct.

**$\beta 2m$  Expression and Purification.** The recombinant human  $\beta 2$  microglobulin ( $h\beta 2m$ ) and murine  $\beta 2$  microglobulin ( $m\beta 2m$ ) were expressed using constructs in pET21d vectors generously supplied by Randall K. Ribaud (32, 33). The constructs were transfected into BL21(DE3) bacteria (Novagen). Cultures were grown at 37 °C to an absorbance of 0.8 at 600 nm and then induced with 1 mM IPTG for 2 h. The bacteria were harvested by centrifugation and washed and resuspended with 0.1 M Tris, pH 8.0, with 2 mM EDTA. Lysozyme (Sigma) was added at 0.5 mg/mL, and the bacteria were incubated overnight at 4 °C. Deoxycholate was added to a final concentration of 0.1%, and the mixture was sonicated on a Brinkmann homogenizer for 4  $\times$  30 s. The inclusion bodies were pelleted by centrifugation at 15000g, and the soluble fraction was discarded. Inclusion bodies were washed three times with 0.1 M Tris, pH 8.0, with 2 mM EDTA and 0.1% deoxycholate followed by a wash with 0.1 M Tris, pH 8.0/2 mM EDTA. The pellet was resuspended with 10 mL of 6 M guanidine hydrochloride/0.1 mM DTT/0.1 M Tris, pH 8.0/2 mM EDTA. The dissolved protein was added to 1 L of prechilled 0.4 M arginine/0.1 M Tris/2 mM EDTA/5 mM reduced glutathione/0.5 mM oxidized glutathione and left in the cold room for a minimum of 3 days to refold. The protein was then diafiltered for 5 volume changes with HBS/3 mM EDTA and then concentrated using an Amicon ultrafiltration cell. The concentrated  $\beta 2m$  was then purified by size exclusion chromatography. Both human and murine  $\beta 2m$  were purified in the same manner.

**Size Exclusion Chromatography.** For  $\beta$ 2m purification, 200  $\mu$ L of protein was loaded onto a Superose 12 column (Pharmacia LKB) using the FPLC system (Pharmacia LKB) and run at 0.4 mL/min with HBS-EP buffer. For analysis and purification of  $\alpha$ 3 domain proteins, 200–500  $\mu$ L of protein was loaded onto a Superdex 75 column (Pharmacia LKB) using the P-500 pump (Pharmacia LKB) at 0.5 mL/min with HBS-EP buffer. Half-milliliter fractions were collected, and protein concentration was determined by 280 nm absorbance. BSA (67 kDa), trypsinogen (24 kDa), and ribonuclease A (13.7 kDa) from Sigma were used as size standards.

**Antibodies and Cell Lines.** Anti-tetra-HIS antibody was purchased from Qiagen. Anti-H-2D<sup>d</sup> conformationally dependent antibody (34.5.8S) was purchased from Pharmingen, San Diego, CA. The LKD8 cell line, a peptide transport deficient EE2H3 embryonic cell line transfected with H-2D<sup>d</sup> (34), was the generous gift of David H. Margulies.

**Flow Cytometry  $\beta$ 2m MHC1 Folding Assay.** Human and murine  $\beta$ 2m were titrated at the indicated concentrations in 12-well tissue culture plates containing 1 mL of OPTI-MEM medium with 0.5% BSA and  $0.5 \times 10^6$  LKD8 cells per well. After an overnight incubation at 26–28 °C, the cells were spun down at 4 °C. A half-microgram of the biotinylated anti-H-2D<sup>d</sup> conformationally dependent antibody, 34.5.8S (Pharmingen), was added to the cells. Cells receiving no primary antibody served as controls. An experiment in which a biotinylated isotype matched antibody (Pharmingen) was used as a control gave similar results. After a 30 min incubation on ice, the cells were washed with 0.5% BSA/OPTI-MEM medium, and 50  $\mu$ L of streptavidin-FITC (Pharmingen) diluted 1:50 in 0.5% BSA/OPTI-MEM was added to the cells for an additional 30 min incubation on ice. The cells were then washed with 3 mL 0.5% BSA/OPTI-MEM and analyzed on a FACScan Plus (Becton and Dickinson, Mountain View, CA) flow cytometer. Data analysis was performed using the Cell Quest software (Becton and Dickinson).

**Surface Plasmon Resonance (SPR) Experiments and Data Analysis.** All SPR experiments were performed on the BIAcore 3000 biosensor (Biacore AB). Anti-HIS antibody, diluted in 10 mM acetate buffer, pH 4.5, was covalently coupled to the carboxymethylated dextran matrix on a CM5 sensor chip (Biacore AB) by using the amine coupling kit as described (35). Experiments were performed in HBS-EP buffer, and regeneration of the anti-HIS surface was achieved with 20 mM HCl.

Equilibrium binding data for  $\beta$ 2m was obtained by averaging a 5–10 s interval of normalized signal after reaching equilibrium. The equilibrium binding data were analyzed by nonlinear curve fitting of the Langmuir isotherm to the data or by linear fitting of a Scatchard plot. Dissociation data was analyzed by 1:1 Langmuir dissociation curve fitting. The curve fitting was performed using the BIAevaluation 3.0 software (Biacore AB). Data were normalized by subtracting the signal for a control surface prior to modeling. Equilibrium and dissociation constants of human and murine  $\beta$ 2m at different temperatures were compared using the Student's *t* test. When multiple comparisons were made, the total *p* value of all significant comparisons was below 0.05. For thermodynamic analysis,  $\Delta G$  was calculated from  $-RT \ln K_a$ , where *R* is the gas

constant, *T* is the temperature in kelvin (K), and  $K_a$  is the equilibrium association constant.  $\Delta H$  and  $\Delta S$  were calculated from slope and intercept of the van't Hoff equation:

$$\ln K_a = -\Delta H/RT + \Delta S/R$$

$\Delta H$  and  $\Delta S$  were also calculated using KaleidaGraph (Synergy Software, Reading, PA) to curve fit a nonlinear equation which takes into account the temperature dependence of  $\Delta H$  and  $\Delta S$  (36):

$$\Delta G^\circ_T = \Delta H_{T_0} - T\Delta S^\circ_{T_0} + \Delta C_p(T - T_0) + T\Delta C_p \ln(T/T_0)$$

where *T* is the temperature in kelvin (K),  $T_0$  is an arbitrary reference temperature (298 K),  $\Delta G^\circ_T$  is the standard free energy of binding (kJ mol<sup>-1</sup>) at *T* and is calculated from the equilibrium constant at temperature *T*,  $\Delta H_{T_0}$  is the enthalpy change (kJ mol<sup>-1</sup>) upon binding at  $T_0$ ,  $\Delta S^\circ_{T_0}$  is the standard state entropy change (kJ mol<sup>-1</sup>) upon binding at  $T_0$ , and  $\Delta C_p$  is the specific heat capacity (kJ mol<sup>-1</sup> K<sup>-1</sup>) and is assumed to be temperature independent. The  $\Delta G$  error was assessed by the SD of determinations from equilibrium binding experiments. The error of  $\Delta H$  and  $\Delta S$  was assessed through the curve fit. In evaluation of the equilibrium constants at differing temperatures, 25 °C (298 K) experiments were generally performed in the same runs to account for non-temperature-dependent variation. The calculated  $\Delta C_p$  was close to zero ( $0.2 \pm 0.8$  kJ mol<sup>-1</sup> K<sup>-1</sup>) as expected from similar values of  $\Delta H$  and  $\Delta S$  seen with linear and nonlinear fits.

**Analytical Ultracentrifugation.** Analytical ultracentrifugation was carried out using a Beckman-Coulter (Fullerton, CA) XL-A centrifuge. An eight-place rotor was used with carbon-filled double-sector centerpiece in the cells. Two of the cells were loaded with different concentrations of SEC-purified  $\alpha$ 3 domain; two were loaded with different concentrations of SEC-purified h $\beta$ 2m; three were loaded with mixtures of different molar ratios of  $\alpha$ 3 and h $\beta$ 2m with total concentrations comparable to those in the other cells. Solution column heights were approximately 5 mm. Rotor speed was 17 000 rpm at an initial temperature of 5 °C because of the time required to attain equilibrium, which was 68 h. The temperature was then raised to 25 °C, but it was found that some of the gradients were not stable at that temperature, probably as a result of the samples being at the higher temperature for 48 h. Thus, only 5 °C results will be presented here.

Data analysis was performed by mathematical modeling using MLAB (Civilized Software, Silver Spring, MD) to fit the data by weighted nonlinear least-squares curve fitting to appropriate models (37). The reactants could be fit with a model for a thermodynamically ideal monomeric solute using the function:

$$A_r = A_b \exp(A'M'(r^2 - r_b^2)) + \epsilon$$

where  $A_r$  is the absorbance at 280 nm as a function of radial position;  $A_b$  is the absorbance at the radial position of the cell bottom  $r_b$  and is a fitting parameter;  $A' = \omega^2/2RT$ , where  $\omega$  is the rotor angular velocity in radians per second, *R* is the gas constant, and *T* is the absolute temperature;  $M'$ , which

is a fitting parameter, is the reduced molar mass and is equal to  $M(1 - \bar{v}\rho)$ , where  $M$  is the anhydrous molar mass,  $\bar{v}$  is the partial specific volume, and  $\rho$  is the solvent density; and  $\epsilon$  is a small baseline error term. The use of this function not only permits assessment of the monomeric state of a component but also experimentally determines the value of the reduced mass of the component in the model for the interaction. For monomeric reactants, this approach is optimal because it eliminates the need of either measuring partial specific volumes or computing them from the amino acid composition and also gives an optimal experimentally determined value for the reduced molar mass. Accordingly, this model was used for globally fitting both of the two appropriate data sets for each component, where  $M'$  was a global parameter and the other fitting parameters were local to each data set. Weight vectors were generated using the EWT function of MLAB. This is done by nonparametrically fitting each data set by multiple smoothing and, by squaring the deviation of each point, obtaining its variance; the individual weight is then the reciprocal of the variance. To obtain meaningful values for the root-mean-square (RMS) error, it is necessary to normalize each weight to unity by multiplying it by the number of values in the weighting vector and dividing this by the sum of the weights in the vector. In effect, this gives a mean value of one for the weights in the vector.

The association constant for the interaction was obtained by using a model for a one-to-one stoichiometry of association and performing global weighted fits to three data sets where the reactants were present in different molar ratios, these being approximately 1:1, 1:2, and 2:1. This model is given by

$$A_r = A_{b,1} \exp(A'M'_1(r^2 - r_b^2)) + A_{b,2} \exp(A'M'_2(r^2 - r_b^2)) + A_{b,1}A_{b,2} \exp(\ln K - \ln E + A'(M'_1 + M'_2)(r^2 - r_b^2)) + \epsilon$$

The  $A_b$ 's are the absorbances for the respective components, the  $A'M$ 's are the reduced masses,  $\ln K$  is the natural logarithm of the molar equilibrium constant, and  $\ln E = \ln(E_1E_2/(E_1 + E_2))$ , where the  $E$ 's are the molar extinction coefficients of the components at 280 nm. This term is necessary to convert  $\ln K$  from a molar scale to an absorbance scale. The fitting parameters are then the global value of  $\ln K$  and the local values of the  $A$ 's and the  $\epsilon$ 's. We fit for  $\ln K$  because this prevents obtaining negative, physically nonmeaningful values for  $K$ , and also, because  $\Delta G^\circ = -RT \ln K$ , we obtain a value with an appropriate error that reflects the change in standard free energy.

## RESULTS

*$\beta$ 2m-Driven Folding of Cell Surface MHC1 Heavy Chain.* To study the species specificity of the  $\beta$ 2m interaction with isolated  $\alpha$ 3 domains, we expressed human and murine  $\beta$ 2m proteins using constructs in bacterial expression vectors (32, 33). To assess the ability of these  $\beta$ 2m molecules to fold MHC1 heavy chain, they were added to a peptide transport deficient cell line, LKD8 (34). This cell line lacks stable folded H-2D<sup>d</sup> heavy chain as demonstrated by poor binding of the conformational epitope dependent anti-H-2D<sup>d</sup> anti-

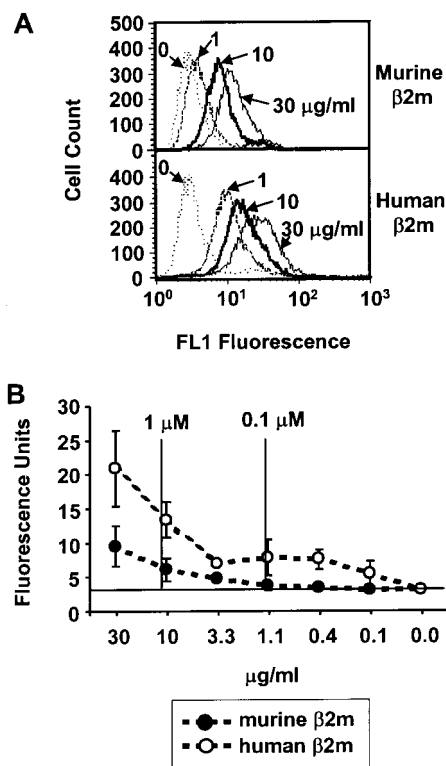


FIGURE 1: Human  $\beta$ 2m is more effective than murine  $\beta$ 2m in refolding of cell surface H-2D<sup>d</sup>. Human or murine  $\beta$ 2m was added overnight at the indicated concentrations to the peptide transport deficient cell line LKD8, and the folding of MHC1 was assessed by evaluating LKD8 staining with a conformationally dependent antibody to H-2D<sup>d</sup> MHC1, 34.5.8S. (A) The staining of the LKD8 cells by biotin-labeled 34.5.8S followed by FITC-streptavidin is increased as the concentration of human or murine  $\beta$ 2m is increased. Histograms of cells incubated with 0 ( $\cdots$ ), 1 ( $---$ ), 10 ( $—$ ), and 30 ( $-$ )  $\mu$ g of  $\beta$ 2m are shown. (B) The median fluorescence of 34.5.8S staining LKD8 cells is increased to a greater extent by human  $\beta$ 2m ( $\circ$ ) than by murine  $\beta$ 2m ( $\bullet$ ). The average and range of two experiments are shown. Controls with FITC-streptavidin alone had median fluorescence of 2–3 units.

body, 34.5.8S. Addition of increasing amounts of either human or murine  $\beta$ 2m to these cells, with an overnight incubation at 26–28 °C, leads to increasing amounts of folded cell surface H-2D<sup>d</sup> MHC1 (Figure 1A). The increase in folded MHC1, shown by the increased binding of 34.5.8S, is greater with the addition of human  $\beta$ 2m than with the addition of murine  $\beta$ 2m (Figure 1B). This confirms previous findings that human  $\beta$ 2m is more effective than murine  $\beta$ 2m in generating folded MHC1 on the surface of peptide transport deficient cell lines (28, 32). Our data, as well as previously published data (20, 21, 28), confirms that with both species of  $\beta$ 2m the increase in folded MHC1 occurs at micromolar concentrations, not the nanomolar concentrations expected from  $\beta$ 2m binding to MHC1 (24–27).

Previous studies have demonstrated that the folding of MHC1 by  $\beta$ 2m, in the absence of added MHC1 binding peptide, is much more efficient at room temperature than at 37 °C (20, 21, 38). This low-temperature induced expression of folded MHC1 in peptide transport deficient cells is dependent on the presence of  $\beta$ 2m. MHC1 molecules expressed in this manner contain very few detectable peptides, as determined by peptide elution and HPLC analysis (39). One possible cause of the low-temperature enhancement of  $\beta$ 2m-induced MHC1 expression is temperature effects on

protein transport to the cell surface. This cannot be the only cause for the temperature dependence since  $\beta$ 2m induced MHC1 at 25–26 °C becomes unstable at 37 °C (21, 38). This occurs in the presence of brefeldin A, so differential protein transport is not a factor in the reduced MHC1 expression. In addition, the folding of MHC1 at 4 °C by  $\beta$ 2m in cell lysates, in which protein transport is not an issue (22, 23), supports  $\beta$ 2m as a critical factor in low-temperature folding of MHC1. Thus, a model for the mechanism of  $\beta$ 2m folding of MHC1 should account for the micromolar concentration, species, and temperature dependence observed.

*Monomeric  $\alpha$ 3 Domains Bind  $\beta$ 2m with a 1:1 Stoichiometry.* To evaluate  $\beta$ 2m binding to isolated  $\alpha$ 3 domains, we needed to verify that such binding could be modeled by a 1:1 Langmuir binding isotherm ( $A + B \rightleftharpoons AB$ ). Size exclusion chromatography (SEC) of the  $\beta$ 2m molecules shows a monomer-sized peak which is purified shortly prior to binding experiments (data not shown). However, the  $\alpha$ 3 domain protein preparations have two size peaks upon SEC (Figure 2A), a predominant peak at approximately 24 kDa and a minor peak at 67 kDa. To confirm that the major peaks of the  $\alpha$ 3 and  $\beta$ 2m proteins are monomeric, we performed analytical ultracentrifugation as per the Experimental Procedures. Using this approach, the reduced molar masses were found to be  $4377 \pm 40$  and  $3462 \pm 121$  for the  $\alpha$ 3 domain and human  $\beta$ 2m, respectively. The weighted RMS errors were 0.0046 and 0.0045 absorbance unit. These reduced masses gave calculated values of 0.2925 and 0.2947 for  $(1 - \bar{v}\rho)$  for the two proteins. These correspond to an approximate value of 0.7 for the partial specific volumes, which is somewhat lower than the values of 0.736 and 0.738 for the respective proteins calculated from their amino acid composition. This difference cannot be attributed to any degree of dimerization of either of the proteins since the quality of the joint fits of a monomeric model to data from significantly different loading concentrations precludes this possibility. Thus, it can be asserted that both proteins are demonstrably monomeric.

To evaluate the binding of  $\beta$ 2m to the  $\alpha$ 3 domain protein, a biosensor for SPR was used. Antibody to the polyhistidine tail of the  $\alpha$ 3 proteins was directly coupled to the carboxymethylated dextran surface of a biosensor chip, allowing for the capture of  $\alpha$ 3 proteins on this surface. Murine  $\beta$ 2m ( $m\beta$ 2m) was injected across the surface of immobilized  $\alpha$ 3 protein, and the binding was assessed by mass-related changes in the sensor chip matrix refractive index and quantified as response units (RU).  $\beta$ 2m was also injected across a control surface consisting of the directly coupled capture antibody without  $\alpha$ 3 protein. This allowed protein concentration effects on the refractive index response to be subtracted out. An example of such an experiment is shown in Figure 2B. The responses of the  $\alpha$ 3 domain and control surfaces were normalized to each other immediately prior to the injection of the  $\beta$ 2m to allow comparison of the active and control surfaces in this figure. Increasing concentrations of  $m\beta$ 2m were then evaluated for binding to either purified monomeric or multimeric  $\alpha$ 3 domains. The binding curves after subtraction of the control surface are shown in Figure 2C. It is clear that the multimer of  $\alpha$ 3 is unable to bind  $\beta$ 2m at the concentrations evaluated, and the  $\alpha$ 3 monomer is responsible for  $\alpha$ 3 binding of  $\beta$ 2m. This was also observed in experiments with human  $\beta$ 2m ( $h\beta$ 2m; data not shown).

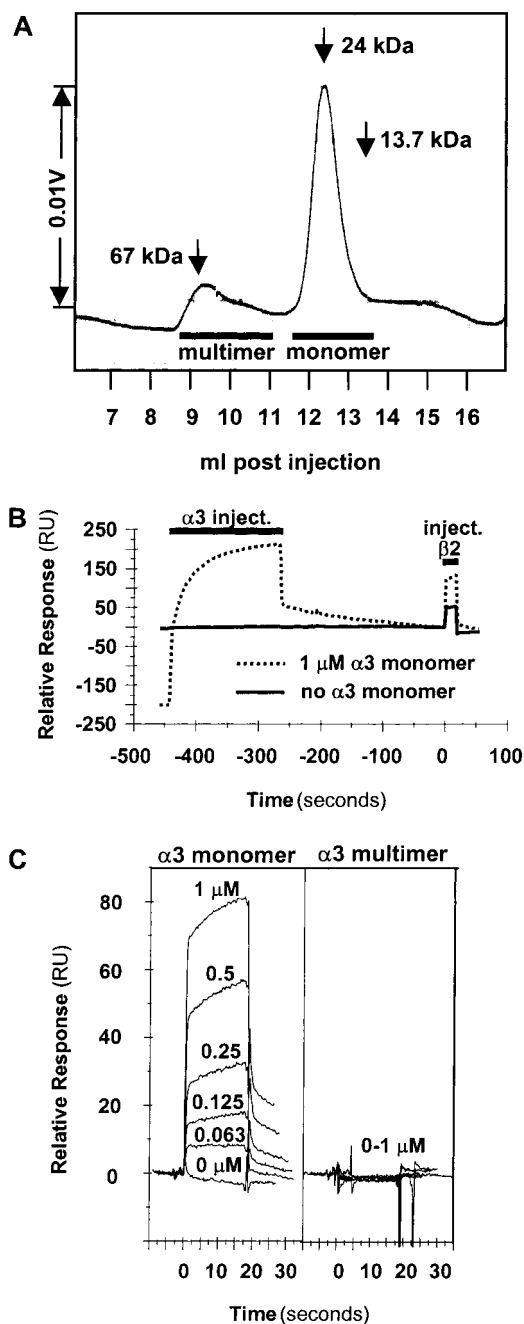


FIGURE 2: Measurement of  $\beta$ 2m binding to the H-2D<sup>d</sup>  $\alpha$ 3 domain. (A) 10  $\mu$ g of  $\alpha$ 3 domain protein was injected on a Superdex 75 size exclusion chromatography column and eluted at 0.5 mL/min with HBS-EP buffer. The protein elution profile at 280 nm absorption is shown with the elution volumes of the standards BSA, trypsinogen, and RNase. Monomeric and multimeric areas of the elution profile are indicated. (B) Anti-polyhistidine antibody was immobilized on two surfaces of an SPR sensor chip at 13 100 and 14 600 RU by amide coupling. 1  $\mu$ M  $\alpha$ 3 domain ( $\cdots$ ) was injected over the 13 100 RU anti-polyhistidine surface, and buffer ( $\text{—}$ ) was injected over the 14 600 RU anti-polyhistidine surface for 3 min at a flow rate of 10  $\mu$ L/min. After a 4.5 min wait, 1  $\mu$ M  $m\beta$ 2m was injected over both surfaces for 20 s at a flow rate of 30  $\mu$ L/min. The responses of the two surfaces were normalized prior to the  $\beta$ 2m injection to allow comparison of the  $\beta$ 2m binding to  $\alpha$ 3 par and the control surface without  $\alpha$ 3 protein. (C) The response of the control surface to  $m\beta$ 2m was subtracted from the response of the  $\alpha$ 3 surface to  $m\beta$ 2m as a measure of  $\beta$ 2m- $\alpha$ 3 binding. Binding is shown for  $\beta$ 2m to surfaces of purified  $\alpha$ 3 monomer and purified  $\alpha$ 3 multimer. The binding curves shown are at the indicated concentrations of  $m\beta$ 2m. Similar results were obtained for comparative binding of  $h\beta$ 2m to  $\alpha$ 3 monomer and  $\alpha$ 3 multimer surfaces.

To evaluate the nature of  $\beta 2m$  binding to monomeric  $\alpha 3$ , we established the stoichiometry of the binding. The addition of an excess of  $\beta 2m$  ( $8 \mu M$ ) to immobilized  $\alpha 3$  and evaluation of the bound RU demonstrated one  $\beta 2m$  per  $\alpha 3$  domain. Excess h $\beta 2m$  led to  $102\% \pm 3\%$  of the expected binding for a 1:1 interaction with the  $\alpha 3$  domain, and excess m $\beta 2m$  led to  $97\% \pm 0.7\%$  of the expected binding for a 1:1 interaction with the  $\alpha 3$  domain. To rule out negative or positive cooperativity between the immobilized  $\alpha 3$  domains upon the binding of  $\beta 2m$ , we evaluated the dependence of  $\beta 2m$  dissociation on the contact time of  $\beta 2m$  with the  $\alpha 3$  domain surface. If the immobilized  $\alpha 3$  domains were not independent in binding  $\beta 2m$ , the increase in contact time would lead to a change in the dissociation curve (40), and such a change with contact time was not observed (data not shown).

To confirm the  $\alpha 3$ - $\beta 2m$  binding with an alternate method, we used analytical ultracentrifugation as described in Experimental Procedures. The global weighted fit of three data sets, with reactants at molar ratios of 1:1, 1:2, and 2:1, gave a value for  $\ln K$  of  $12.47 \pm 0.16$ , which gives a value for  $\Delta G^\circ$  of  $-28.8 \pm 0.38$  kJ/mol ( $-6.89 \pm 0.09$  kcal/mol) at 5 °C and a value of  $3.86 \pm 0.62 \mu M$  for the equilibrium dissociation constant. The joint fit had a weighted RMS error of 0.0038 absorbance unit, clearly indicating that the one-to-one stoichiometry was the correct model for the association. The equilibrium dissociation constant, although higher than that measured using the Biacore, is still in the micromolar range. Biacore measurements of the interaction immobilizing  $\beta 2m$  also gave a somewhat higher dissociation constant than that of the interaction with immobilized  $\alpha 3$  (data not shown). The  $\alpha 3$  peptide extension and polyhistidine tail may interfere with the binding and are sequestered when  $\alpha 3$  is immobilized through antibody to the polyhistidine tail.

Our data demonstrate that the  $\alpha 3$  domain exists as a monomer, it binds  $\beta 2m$  with a 1:1 stoichiometry, and this binding shows no evidence of cooperativity. The  $\beta 2m$  interaction with the  $\alpha 3$  domain can be modeled by 1:1 Langmuir binding and occurs with a dissociation in the micromolar range.

**$\beta 2m$  Equilibrium Binding to the  $\alpha 3$  Domain.** Human and murine  $\beta 2m$ -microglobulin were used to evaluate the binding parameters of  $\beta 2m$  to the MHC1  $\alpha 3$  domain. Whole murine MHC class I binds human  $\beta 2m$  with a  $K_d$  of  $4 \times 10^{-10}$  M (27) and murine  $\beta 2m$  with a 3-fold lower affinity (41). We evaluated  $\beta 2m$  binding by comparing  $\alpha 3$  parental ( $\alpha 3$  par) and control surfaces as shown in Figure 2B and taking equilibrium values of the subtracted data as shown in Figure 2C. The equilibrium values were used to generate the plots in Figure 3, which were used for equilibrium fitting. Figure 3A shows nonlinear curve fits of the Langmuir isotherm for murine and human  $\beta 2m$  binding to the  $\alpha 3$  domain. The same data are graphed as Scatchard plots in Figure 3B. The linearity of the Scatchard plots confirms the lack of cooperativity in the binding.

The equilibrium constants generated from three experiments with each of the  $\beta 2m$  proteins are shown in Table 1. Human  $\beta 2m$  binds isolated  $\alpha 3$  domains with an equilibrium  $K_d$  of  $3.4 \times 10^{-7}$  M, which is 2-fold better than murine  $\beta 2m$ . This advantage of human  $\beta 2m$  remains when the  $\alpha 3$  CD loop is mutated at position 227 with concurrent loss of CD8 binding (29). The superior binding of human  $\beta 2m$  to isolated

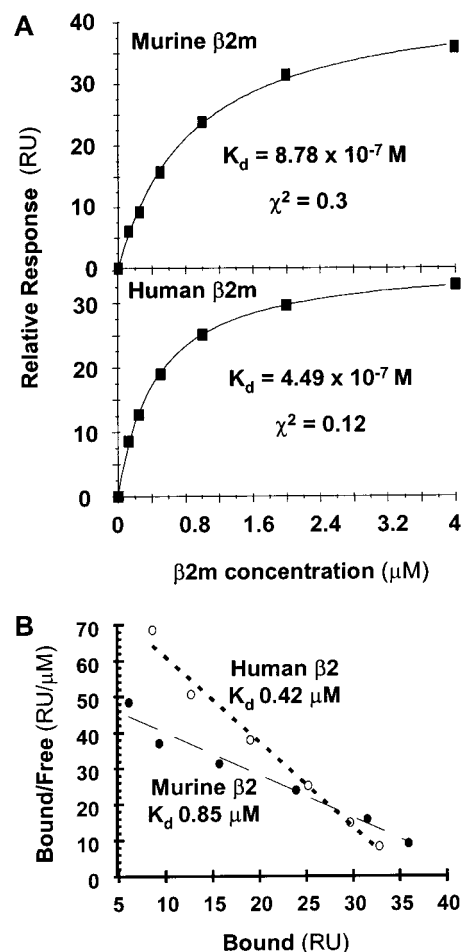


FIGURE 3: Human  $\beta 2m$  has a higher affinity for the  $\alpha 3$  domain than murine  $\beta 2m$ . (A) Averages of the equilibrium  $\beta 2m$ - $\alpha 3$  binding responses during  $\beta 2m$  injection were plotted versus  $\beta 2m$  concentration. These points were fit with the Langmuir isotherm to determine the  $K_d$ . The  $\chi^2$  (average of the squared residuals) values of 0.3 and 0.12 suggest good fits. (B) The same equilibrium binding data for human (○) and murine (●)  $\beta 2m$  is shown transformed into a Scatchard plot. The  $K_d$  was determined from the negative inverse of the slope of the fitted line.

$\alpha 3$  domains correlates with the superior MHC1 folding ability previously shown. In Table 1 we also evaluate the effect of temperature on  $\beta 2m$  binding to the  $\alpha 3$  domain. The affinity of both human and murine  $\beta 2m$  is decreased 3-fold by raising the assay temperature from 25 to 37 °C. The improved  $\beta 2m$  binding to the  $\alpha 3$  domain at room temperature correlates with the superior  $\beta 2m$ -driven refolding of MHC1 at room temperature (20, 21, 38). It is of note that the difference in binding between human and murine  $\beta 2m$  to  $\alpha 3$  domains is only 2-fold while human  $\beta 2m$  has been reported to be 4–20 times more effective at folding MHC1 heavy chain than murine  $\beta 2m$  (28). One possible explanation for this difference is that h $\beta 2m$  has an advantage in refolding  $\alpha 1\alpha 2$  domains in addition to binding the  $\alpha 3$  domain. The MHC1 refolding advantage of h $\beta 2m$  was reduced by mutation of a  $\beta 2m$  residue which contacts the  $\alpha 1\alpha 2$  domains of MHC1 (32). Since mutations in the  $\alpha 1\alpha 2$  interface of  $\beta 2m$  have variable effects on the folding of different MHC1 haplotypes, this interface may also contribute to the larger overall advantage of human versus murine  $\beta 2m$ .

There is another factor that could explain the difference between relative  $\alpha 3$  binding and relative MHC1 refolding

Table 1: Equilibrium Binding of  $\beta$ 2m to MHC1  $\alpha$ 3 Domain Proteins<sup>a</sup>

$\beta$ 2m protein	$\alpha$ 3 protein	25 °C		37 °C	
		$K_d$ (M)	$\chi^2$	$K_d$ (M)	$\chi^2$
murine	$\alpha$ 3 par	$6.9 \pm 1.7 \times 10^{-7}$	$0.29 \pm 0.21$	$2.1 \pm 0.27 \times 10^{-6}$	$0.18 \pm 0.11$
murine	$\alpha$ 3 227m	$6.0 \pm 1.3 \times 10^{-7}$	$0.13 \pm 0.03$	not performed	
human	$\alpha$ 3 par	$3.4 \pm 1.1 \times 10^{-7}$	$0.11 \pm 0.04$	$1.1 \pm 0.19 \times 10^{-6}$	$0.09 \pm 0.05$
human	$\alpha$ 3 227m	$3.9 \pm 0.3 \times 10^{-7}$	$0.14 \pm 0.07$	not performed	

<sup>a</sup> The equilibrium fit was determined from a nonlinear Langmuir isotherm fit of equilibrium data similar to that of Figure 3A. The fitting was done using the Biaevaluation 3.0 software as per the Experimental Procedures. The  $K_d$  results are the average of three experiments  $\pm$  the standard deviation. The  $\chi^2$  average  $\pm$  the standard deviation is shown as a measure of the curve fit. The  $K_d$  of both  $\beta$ 2m are significantly different from each other ( $p < 0.045$ ). The  $K_d$  at 37 °C is significantly different from the  $K_d$  at 25 °C ( $p < 0.002$ ).

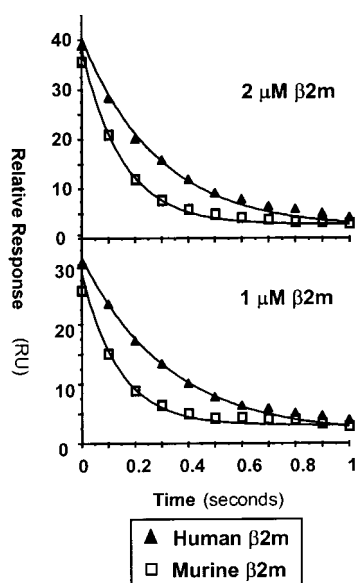


FIGURE 4: Human  $\beta$ 2m dissociates more slowly from the  $\alpha$ 3 domain than murine  $\beta$ 2m. The dissociation of human ( $\blacktriangle$ ) and murine ( $\square$ )  $\beta$ 2m from  $\alpha$ 3 is shown after the injection of 1 or 2  $\mu$ M  $\beta$ 2m. The curves are 1:1 Langmuir fits of the dissociation data used in determining the dissociation constants shown in Table 2.

of the different  $\beta$ 2m species. After  $\beta$ 2m binding of the MHC1  $\alpha$ 3 domain, a discrete interval of time may be required for potential multiple steps in the  $\beta$ 2m folding of the  $\alpha$ 1 $\alpha$ 2 domains of the MHC1 heavy chain. If this is the case, the association of  $\beta$ 2m with the MHC1  $\alpha$ 3 domain may be a kinetic intermediate, and a 2-fold change in half-life of that intermediate may have greater than a 2-fold effect on the yield of folded MHC1. This greater effect would operate through a mechanism similar to kinetic proofreading (42). The shorter murine  $\beta$ 2m dwell time would decrease the probability of an initial step in  $\alpha$ 1 $\alpha$ 2 folding that would slow the  $\beta$ 2m dissociation. This would then lead to an even lower probability of completion of  $\alpha$ 1 $\alpha$ 2 domain folding than the difference in  $\beta$ 2m- $\alpha$ 3 half-life. To explore this possibility, we evaluated the dissociation kinetics of human and murine  $\beta$ 2m binding to the  $\alpha$ 3 domain.

*Human  $\beta$ 2m Bound to the  $\alpha$ 3 Domain Has a 2-fold Longer Half-Life Than Murine  $\beta$ 2m.* To evaluate the dissociation of the  $\beta$ 2m from the  $\alpha$ 3 domain of MHC1,  $\alpha$ 3 was immobilized on a sensor chip by capture with anti-HIS antibody, and  $\beta$ 2m binding was evaluated as described above. Because of the rapidity of the dissociation, data points were captured at 0.1 s intervals and a high flow rate of 30  $\mu$ L/min was used for the  $\beta$ 2m injection. Figure 4 shows the dissociation phase of the  $\beta$ 2m- $\alpha$ 3 interaction and demonstrates that human  $\beta$ 2m dissociates more slowly than murine  $\beta$ 2m. Table 2 shows

Table 2: Dissociation of Human and Murine  $\beta$ 2m from MHC1  $\alpha$ 3 Domain Protein<sup>a</sup>

$\beta$ 2m concn ( $\mu$ M)	$k_{off}$ ( $s^{-1}$ )	
	human $\beta$ 2m	murine $\beta$ 2m
4	2.42	3.18
2.5	2.48	5.78
2	1.49	4.30
1.25	2.72	4.27
1	1.54	4.01
0.625	2.45	4.17
av $\pm$ SD	$2.18 \pm 0.53$	$4.29 \pm 0.84$

<sup>a</sup> The dissociation constants ( $k_{off}$ s) were determined from a nonlinear Langmuir isotherm fit of dissociation data similar to that of Figure 4. The fitting was done using the Biaevaluation 3.0 software as per the Experimental Procedures. The  $k_{off}$  results at different concentrations of  $\beta$ 2m were averaged, and the standard deviation is shown. The average  $\chi^2$  for the h $\beta$ 2m fits was 0.2 and for the m $\beta$ 2m fits was 0.69. The dissociation constants for human and murine  $\beta$ 2m are significantly different from each other ( $p < 0.01$ ).

the dissociation constants of human and murine  $\beta$ 2m at a number of concentrations. No concentration dependence is evident, and the off rate of human  $\beta$ 2m is one-half that of murine  $\beta$ 2m. If the binding of  $\beta$ 2m to the  $\alpha$ 3 domain of MHC1 is a kinetic intermediate in the  $\beta$ 2m folding of MHC1, then the 2-fold longer dwell time of human  $\beta$ 2m could account for an even greater effect on MHC1 folding.

*$\beta$ 2m Binding to the  $\alpha$ 3 Domain Is Associated with Decreased Entropy.* The temperature dependence of  $\beta$ 2m refolding of cell surface MHC1 and binding to the MHC1  $\alpha$ 3 domain is suggestive of an enthalpic-driven process in which entropy decreases upon binding. To better define the thermodynamic parameters of  $\beta$ 2m- $\alpha$ 3 binding, we evaluated the equilibrium binding of murine  $\beta$ 2m to  $\alpha$ 3 at a number of different temperatures. We evaluated these SPR data as described by Roos et al. (43) using both a linear and nonlinear van't Hoff analysis. Figure 5A shows the plot of the linear analysis and the values of  $\Delta H$  obtained from the slope and  $-T\Delta S$  derived from  $\Delta H$  and  $\Delta G$ . The nonlinear analysis leads to very similar results with a  $\Delta G$  of  $-35.2$ , a  $\Delta H$  of  $-70.7$ , and a  $-T\Delta S$  of  $35.6$  kJ/mol at 25 °C. In both cases  $-T\Delta S$  has a positive value, suggesting a reduction of disorder during  $\beta$ 2m- $\alpha$ 3 binding. This is an unusual binding thermodynamics as shown by comparing the values to those of 30 nonantibody protein-protein interactions (44) (Figure 5B). Binding with decreasing disorder can be seen with TCR-MHC-peptide interactions and some antibody-antigen interactions (45, 46).

## DISCUSSION

$\beta$ 2m is both a structural component of MHC1 and a chaperone for correct folding of the MHC1 heavy chain.  $\beta$ 2m

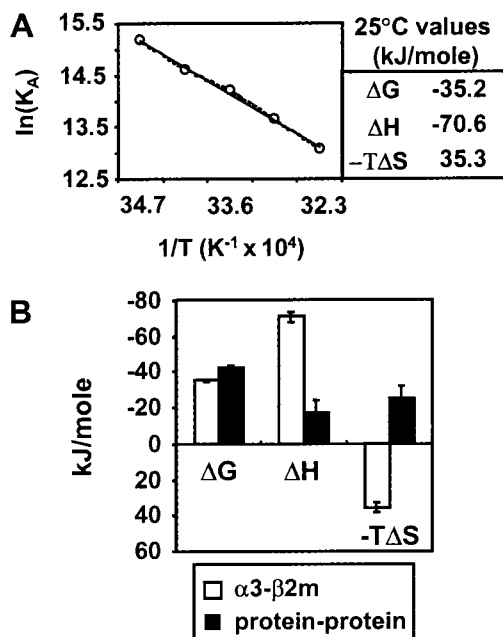


FIGURE 5:  $\beta 2\text{m}$  binding to the  $\alpha 3$  domain has an unfavorable entropy. (A) The equilibrium association binding constant ( $K_A$ ) of  $\beta 2\text{m}$  to  $\alpha 3$  was determined at the temperatures 15, 20, 25, 30, and 37 °C (two to eight evaluations per temperature). The natural logarithm of the  $K_A$  was graphed against the inverse of temperature using the van't Hoff equation. The values of  $\Delta H$  were obtained from the slope of the line, and  $-T\Delta S$  was derived from  $\Delta H$  and  $\Delta G$ . (B) The thermodynamic parameters determined from the nonlinear van't Hoff relationship for  $\beta 2\text{m}$ – $\alpha 3$  binding (□) were compared to those of a series of 30 nonantibody protein–protein interactions (■). The error bars for  $\beta 2\text{m}$ – $\alpha 3$  binding are based on the SD of measured values or the error of curve fitting as per the Experimental Procedures. The error bars for the protein–protein interactions are the SEM of reported values.

binds folded MHC1 heavy chain at a 1000-fold lower concentration than that required for  $\beta 2\text{m}$  refolding of cell surface MHC1. The  $\alpha 3$  domain of the MHC1 heavy chain is folded before the other MHC1 domains prior to associating with  $\beta 2\text{m}$  (30, 31) and remains folded longest on the cell surface (21). This makes a  $\beta 2\text{m}$  interaction with the  $\alpha 3$  domain of heavy chain an attractive model for the  $\beta 2\text{m}$  refolding of MHC1 heavy chain. We have previously demonstrated that human  $\beta 2\text{m}$  binding to an isolated  $\alpha 3$  domain has a micromolar affinity (29) similar to the concentration that  $\beta 2\text{m}$  refolds MHC1. To provide further evidence that  $\beta 2\text{m}$  refolds MHC1 by binding the  $\alpha 3$  domain, we demonstrated that the  $\alpha 3$  domain binds murine  $\beta 2\text{m}$  in addition to binding human  $\beta 2\text{m}$ . The binding of the  $\alpha 3$  domain by  $\beta 2\text{m}$  has a species hierarchy similar to the refolding of the MHC1 heavy chain. This  $\beta 2\text{m}$  species difference in affinity is due to a difference in off rates and may be important in determining the half-life of a kinetic intermediate in MHC1 folding. We have also demonstrated that the  $\beta 2\text{m}$ – $\alpha 3$  interaction has the same unusual temperature dependence as  $\beta 2\text{m}$ -induced refolding of MHC1. These data strongly support a model in which a  $\beta 2\text{m}$ – $\alpha 3$  interaction facilitates the folding of the MHC1 heavy chain.

A crystal structure of  $\beta 2\text{m}$  binding the  $\alpha 1\alpha 2$  domains of MHC1 (47) suggests that  $\beta 2\text{m}$  can bind MHC1 in the absence of the  $\alpha 3$  domain. It is of note that the MHC1  $\alpha 3$  domain was present during the refolding in this study and was only lost due to proteolysis during crystallization. MHC1

without the  $\alpha 3$  domain can be expressed on cells and folded by large amounts of high-affinity peptide; however, the  $\alpha 3$ -deleted MHC1 is unable to be folded by  $\beta 2\text{m}$  at any tested concentration (23). Therefore, neither of these studies demonstrates  $\beta 2\text{m}$ -driven MHC folding without the  $\alpha 3$  domain.

The temperature dependence of the  $\beta 2\text{m}$ – $\alpha 3$  interaction led us to evaluate the thermodynamics of the interaction. We found that the binding is driven by a large enthalpy contribution which overcomes an unfavorable entropy of binding. This type of binding is generally not seen in protein–protein interactions which have coevolved to favor binding. Recently, binding which decreases disorder has been seen in T-cell receptor (TCR) interactions with peptide–MHC ligand (45, 46). TCR does not necessarily coevolve with potential antigenic peptides, and the basic TCR structure must recognize a wide range of peptide–MHC complexes. The cross-reactivity of TCR recognition supports a model in which a flexible TCR can scan through a number of conformations until it optimizes interaction with a specific ligand interaction. The limitation of this flexibility upon binding the ligand increases order and exacts an entropic penalty on the overall binding affinity. Although the relatively nonpolymorphic  $\beta 2\text{m}$  must interact with a wide variety of polymorphic MHC1 chains,  $\beta 2\text{m}$  has coevolved with MHC1, and MHC1 polymorphisms tend to involve the peptide and TCR contacts. Scanning through a series of conformations is unlikely to be the cause of the increased order observed in the  $\beta 2\text{m}$ – $\alpha 3$  interaction.  $\beta 2\text{m}$  can fold the complete MHC1 heavy chain, and one previous study suggested that there is an entropic penalty for  $\beta 2\text{m}$  exchange with whole MHC1 (25). The increased order involved in folding MHC1 could account for an unfavorable entropy of binding. This folding may also involve conformational changes in the  $\alpha 3$  domain or  $\beta 2\text{m}$  and account for our observed binding thermodynamics. Modeling of  $\beta 2\text{m}$  interactions with MHC1 heavy chains suggests a conformational change of  $\beta 2\text{m}$  upon binding (48). In addition, although we have been unable to accurately model the association constant in our antibody capture binding system, a number of experiments suggest an association constant of  $10^6 \text{ M}^{-1} \text{ s}^{-1}$ , which is similar to the association constant calculated from the observed equilibrium and dissociation constants. This rapid on rate is different than the slow on rates seen with the TCR interactions and suggests that the entropic penalty seen during the  $\beta 2\text{m}$ – $\alpha 3$  interaction is due to a defined conformational shift or ordering of water molecules rather than the time-consuming scanning of multiple conformations. Another example of binding with an increase in order is seen with Z, a one-domain analogue of protein A (43). This binding also has a rapid association constant in the range of  $10^6 \text{ M}^{-1} \text{ s}^{-1}$ . This binding may also require a defined conformational change upon binding.

The binding and thermodynamic parameters of  $\beta 2\text{m}$  interactions play an important role in MHC1 assembly in the ER. The  $\beta 2\text{m}$  interaction with the  $\alpha 3$  domain can be further evaluated with other fragments of the MHC1 heavy chain and whole MHC1 using SPR. The addition of MHC binding peptides and chaperone molecules such as calreticulin, tapasin, TAP, and Erp57 may allow for the understanding of the energetics of physiologic MHC1 folding in the ER.



The refolding of the free MHC1 heavy chain on the surface of cells by  $\beta$ 2m plays an important role in normal cells (49–51) as well as peptide transport deficient cells. This refolding can have immunologic consequences due to the enhanced presentation of exogenous peptide fragments to CTL (52–55). Such presentation of extracellularly processed (56–59) class I restricted peptides could cause lysis of uninfected bystander cells in a CTL response to virus. Understanding the mechanism of MHC1 refolding may suggest strategies to prevent such aberrant responses.

The binding of the MHC1 heavy chain to  $\beta$ 2m involves multiple domains with multiple contacts. We have analyzed the binding of isolated MHC1  $\alpha$ 3 domains to  $\beta$ 2m using SPR and obtained information important for understanding the mechanism of MHC1 folding. The analysis of the binding characteristics of individual domains and surfaces of complex biologic molecules is an important tool in elucidating the structural and functional behavior of these molecules.

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## REFERENCES

1. Townsend, A., and Bodmer, H. (1989) *Annu. Rev. Immunol.* 7, 601–624.
2. Jones, E. Y., Tormo, J., Reid, S. W., and Stuart, D. I. (1998) *Immunol. Rev.* 163, 121–128.
3. Germain, R. N., and Margulies, D. H. (1993) *Annu. Rev. Immunol.* 11, 403–450.
4. Ljunggren, H. G., and Thorpe, C. J. (1996) *Histol. Histopathol.* 11, 267–274.
5. Ozato, K., Evans, G. A., Shykind, B., Margulies, D. H., and Seidman, J. G. (1983) *Proc. Natl. Acad. Sci. U.S.A.* 80, 2040–2043.
6. Bjorkman, P. J., Saper, M. A., Samraoui, B., Bennett, W. S., Strominger, J. L., and Wiley, D. C. (1987) *Nature* 329, 512–518.
7. Davis, M. M., Boniface, J. J., Reich, Z., Lyons, D., Hampl, J., Arden, B., and Chien, Y. (1998) *Annu. Rev. Immunol.* 16, 523–544.
8. Saper, M. A., Bjorkman, P. J., and Wiley, D. C. (1991) *J. Mol. Biol.* 219, 277–319.
9. Santos-Aguado, J., Biro, P. A., Fuhrmann, U., Strominger, J. L., and Barbosa, J. A. (1987) *Mol. Cell. Biol.* 7, 982–990.
10. Ribaldo, R. K., and Margulies, D. H. (1995) *J. Immunol.* 155, 3481–3493.
11. Salter, R. D. (1992) *Hum. Immunol.* 35, 40–49.
12. Seong, R. H., Clayberger, C. A., Krensky, A. M., and Parnes, J. R. (1988) *J. Exp. Med.* 167, 288–299.
13. Zijlstra, M., Bix, M., Simister, N. E., Loring, J. M., Raulet, D. H., and Jaenisch, R. (1990) *Nature* 344, 742–746.
14. Allen, H., Fraser, J., Flyer, D., Calvin, S., and Flavell, R. (1986) *Proc. Natl. Acad. Sci. U.S.A.* 83, 7447–7451.
15. Bix, M., and Raulet, D. (1992) *J. Exp. Med.* 176, 829–834.
16. Solheim, J. C., Johnson, N. A., Carreno, B. M., Lie, W. R., and Hansen, T. H. (1995) *Eur. J. Immunol.* 25, 3011–3016.
17. Fink, A. L. (1999) *Physiol. Rev.* 79, 425–449.
18. Solheim, J. C. (1999) *Immunol. Rev.* 172, 11–19.
19. Cresswell, P., Bangia, N., Dick, T., and Diedrich, G. (1999) *Immunol. Rev.* 172, 21–28.
20. Rock, K. L., Gramm, C., and Benacerraf, B. (1991) *Proc. Natl. Acad. Sci. U.S.A.* 88, 4200–4204.
21. Otten, G. R., Bikoff, E., Ribaldo, R. K., Kozlowski, S., Margulies, D. H., and Germain, R. N. (1992) *J. Immunol.* 148, 3723–3732.
22. Elliott, T., Cerundolo, V., Elvin, J., and Townsend, A. (1991) *Nature* 351, 402–406.
23. Elliott, T., Elvin, J., Cerundolo, V., Allen, H., and Townsend, A. (1992) *Eur. J. Immunol.* 22, 2085–2091.
24. Hyafil, F., and Strominger, J. L. (1979) *Proc. Natl. Acad. Sci. U.S.A.* 76, 5834–5838.
25. Parker, K. C., and Strominger, J. L. (1985) *Biochemistry* 24, 5543–5550.
26. Hochman, J. H., Shimizu, Y., DeMars, R., and Edidin, M. (1988) *J. Immunol.* 140, 2322–2329.
27. Pedersen, L. O., Hansen, A. S., Olsen, A. C., Gerwien, J., Nissen, M. H., and Buus, S. (1994) *Scand. J. Immunol.* 39, 64–72.
28. Shields, M. J., Moffat, L. E., and Ribaldo, R. K. (1998) *Mol. Immunol.* 35, 919–928.
29. Whitman, M. C., Strohmaier, J., O’Boyle, K., Tingem, J. M., Wilkinson, Y., Goldstein, J., Chen, T., Brorson, K., Brunswick, M., and Kozlowski, S. (2000) *Mol. Immunol.* 37, 141–149.
30. Elliott, T. (1991) *Immunol. Today* 12, 386–388.
31. Ribaldo, R. K., and Margulies, D. H. (1992) *J. Immunol.* 149, 2935–2944.
32. Shields, M. J., Assefi, N., Hodgson, W., Kim, E. J., and Ribaldo, R. K. (1998) *J. Immunol.* 160, 2297–2307.
33. Shields, M. J., Kubota, R., Hodgson, W., Jacobson, S., Biddison, W. E., and Ribaldo, R. K. (1998) *J. Biol. Chem.* 273, 28010–28018.
34. Bikoff, E. K., Otten, G. R., and Robertson, E. J. (1991) *Eur. J. Immunol.* 21, 1997–2004.
35. Johnsson, B., Lofas, S., and Lindquist, G. (1991) *Anal. Biochem.* 198, 268–277.
36. Naghibi, H., Tamura, A., and Sturtevant, J. M. (1995) *Proc. Natl. Acad. Sci. U.S.A.* 92, 5597–5599.
37. Lewis, M. S. (1996) in *Exploration* 3, pp 11–16, Beckman-Coulter.
38. Ljunggren, H. G., Stam, N. J., Ohlen, C., Neeffjes, J. J., Hoglund, P., Heemels, M. T., Bastin, J., Schumacher, T. N., Townsend, A., Karre, K., et al. (1990) *Nature* 346, 476–480.
39. Anderson, K. S., Alexander, J., Wei, M., and Cresswell, P. (1993) *J. Immunol.* 151, 3407–3419.
40. Karlsson, R., and Falt, A. (1997) *J. Immunol. Methods* 200, 121–133.
41. Pedersen, L. O., Stryhn, A., Holter, T. L., Etzerodt, M., Gerwien, J., Nissen, M. H., Thogersen, H. C., and Buus, S. (1995) *Eur. J. Immunol.* 25, 1609–1616.
42. Hopefield, J. J. (1974) *Proc. Natl. Acad. Sci. U.S.A.* 71, 4135–4139.
43. Roos, H., Karlsson, R., Nilshans, H., and Persson, A. (1998) *J. Mol. Recognit.* 11, 204–210.
44. Stites, W. E. (1997) *Chem. Rev.* 97, 1233–1250.
45. Willcox, B. E., Gao, G. F., Wyer, J. R., Ladbury, J. E., Bell, J. I., Jakobsen, B. K., and van der Merwe, P. A. (1999) *Immunity* 10, 357–365.
46. Boniface, J. J., Reich, Z., Lyons, D. S., and Davis, M. M. (1999) *Proc. Natl. Acad. Sci. U.S.A.* 96, 11446–11451.
47. Collins, E. J., Garboczi, D. N., Karpusas, M. N., and Wiley, D. C. (1995) *Proc. Natl. Acad. Sci. U.S.A.* 92, 1218–1221.
48. Shields, M. J., Hodgson, W., and Ribaldo, R. K. (1999) *Mol. Immunol.* 36, 561–573.
49. Smith, M. H., and Barber, B. H. (1990) *Mol. Immunol.* 27, 169–180.
50. Rock, K. L., Gamble, S., Rothstein, L., Gramm, C., and Benacerraf, B. (1991) *Cell* 65, 611–620.
51. Smith, J. D., Lie, W. R., Gorka, J., Myers, N. B., and Hansen, T. H. (1992) *Proc. Natl. Acad. Sci. U.S.A.* 89, 7767–7771.
52. Vitiello, A., Potter, T. A., and Sherman, L. A. (1990) *Science* 250, 1423–1426.
53. Rock, K. L., Rothstein, L. E., Gamble, S. R., and Benacerraf, B. (1990) *Proc. Natl. Acad. Sci. U.S.A.* 87, 7517–7521.
54. Kozlowski, S., Takeshita, T., Boehncke, W. H., Takahashi, H., Boyd, L. F., Germain, R. N., Berzofsky, J. A., and Margulies, D. H. (1991) *Nature* 349, 74–77.

55. Kane, K. P., Sherman, L. A., and Mescher, M. F. (1991) *Eur. J. Immunol.* 21, 2289–2292.
56. Sherman, L. A., Burke, T. A., and Biggs, J. A. (1992) *J. Exp. Med.* 175, 1221–1226.
57. Kozlowski, S., Corr, M., Takeshita, T., Boyd, L. F., Pendleton, C. D., Germain, R. N., Berzofsky, J. A., and Margulies, D. H. (1992) *J. Exp. Med.* 175, 1417–1422.
58. Falco, L. D., Jr., Colarusso, L. J., Benacerraf, B., and Rock, K. L. (1992) *Proc. Natl. Acad. Sci. U.S.A.* 89, 8347–8350.
59. Kozlowski, S., Corr, M., Shirai, M., Boyd, L. F., Pendleton, C. D., Berzofsky, J. A., and Margulies, D. H. (1993) *J. Immunol.* 151, 4033–4044.

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